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2.19 A solution containing 30g of non-volatile solute exactly in 90 g of water has a vapour pressure of 2.8 kPa at 298 K. Further, 18g of water is then added to the solution and the new of vapour pressure becomes 2.9 kPa at 298 K. Calculate (i) molar mass of the solute. (ii) vapour pressure of water at 298 K. Sol. Let the molar mass of solute ...

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### **Answers to Chapters 1,2,3,4,5,6,7,8,9 - End of Chapter ...**

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From Chapter 21 on, I switched to Visual Studio Community 2013. Chapter 1 has no programming exercises, and for Chapter 2 I just went along and changed the same file over and over, so these solutions are not here. Chapters 12-16 deal with GUI/graphics and use FLTK.

### **GitHub - bewuethr/stroustrup-ppp: Exercises from ...**

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