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CHAPTER 4 TEST: Atoms, Atomic Theory and Atomic Structure Matching. A. Bohr B. Democritus C. Rutherford D. Dalton E. Thomson F. Schrodinger ____ 1. Greek thinker; called nature's basic particle an atom, based on the Greek word "atomos" which means "indivisible". Did not have evidence that atoms existed. ____ 2. <http://www.brooklyn.k12.oh.us/userfiles/75/Classes/4711/NON-HONORS%20Chapter%204%20Test%202010.pdf>.

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86 Chapter 4 What You'll Learn You will identify the experi-ments that led to the devel-opment of the nuclear model of atomic structure. You will describe the structure of the atom and differ-entiate among the subatomic particles that comprise it. You will explain the rela-tionship between nuclear stability and radioactivity. Why It's important

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PPT - Review Chapter 4 Atomic Structure Test: Wednesday ...
Chapter 4 "Atomic Structure" Tools. Copy this to my account; E-mail to a friend; ... Help; Use these activities to help your study the vocabulary terms from this chapter. A B; atomic mass unit: defined as the mass of a carbon -12 atom; nucleus: central core of an atom, which contains most of the atom's mass; group: a vertical column of elements ...